

Course Code	Course Title	C	H	I	E	T
17P3CMC8	INORGANIC CHEMISTRY-III	4	4	25	75	100

UNIT I INTRODUCTION TO CO-ORDINATION CHEMISTRY 12 Hrs

Introduction – IUPAC nomenclature- co-ordination numbers- Isomerism- structural, geometrical and optical isomerism in square planar and octahedral complexes. Chelating ligand, chelate effect and its application, stability of complexes – complex equilibria – determination of stability constant- Spectrophotometric- Job's variation - Potentiometric methods. factors affecting stability constants. Irving-William series.

UNIT II THEORY OF CO-ORDINATION COMPOUNDS. 12 Hrs

Valence bond theory. Crystal field splittings in octahedral, tetrahedral, square pyramidal, trigonal bipyramidal and square planar geometries-Consequences of crystal field splitting on thermo dynamic, magnetic and spectroscopic properties. Applications of CFT, spectrochemical series-Nephelauxetic effect. Jahn-Teller effect. Inadequacy of CFT. The molecular orbital theory-octahedral, tetrahedral and square planar geometries. Experimental evidences for covalence in metal-ligand bonding.

UNIT III ELECTRONIC SPECTRA OF COMPLEX 12 Hrs

Atomic Term Symbol: Ground and higher states — methods of determining ground state term and complete terms –Pigeon hole diagram and Russel-Saunders microstates method for p^2 and d^2 configuration - term symbols for non- equivalent electrons.

A detailed survey of the electronic spectra of octahedral, tetrahedral and square planar complexes of transition metals-(i) effect of spin orbit coupling ii) Jahn-Teller distortions and its effect on the structure and spectrum, the nephelauxetic effect and the nephelauxetic series-Calculation of Dq, B, C from the spectrum. Orgel and Tanabe-Sugano diagrams and evaluation of $10Dq$ and β for octahedral complexes of d^3, d^8 and tetrahedral complexes of d^2, d^7 configurations. Charge transfer spectra of complexes-ligand to metal, metal to ligand types and metal to metal type.

UNIT IV NUCLEAR MAGNETIC RESONANCE 12 Hrs

$^1H, ^{13}C, ^{19}F$ and ^{31}P – NMR. Applications to inorganic compounds - NMR of paramagnetic complexes. Contact and pseudo contact interactions- structures of metal complexes-shift reagents structural problems of the compounds like $CFCl_3, SF_4, PF_5, BrF_5, H_2PF_3, PF_3(NH_2)_2, P_4S_3$, *mer* and *fac*- $Rh(PPh_3)_3Cl_3$, *cis* and *trans*- $[PtCl_2(PBu_3)(PPh_3)_3]$. Fluxional behavior of complexes.

Electron spin resonance: Spin-Hamiltonian, ESR phenomenon - hyperfine interactions-spin-orbit coupling- ESR spectra of D and F states in octahedral fields (Zero field splitting and Kramers degeneracy)- determination of the ground states of complexes. EPR spectra of transition metal ions complexes like bis(salicylaldehyde)copper(II), $Co_3(CO)_9Se$, $[(NH_3)_5Co-O_2-$

Co(NH₃)₅]⁵⁺, NH₂ radical and Mn²⁺ complexes – representative spectra of different dⁿ systems – evaluation of spin – orbit coupling.

UNIT V PHOTOELECTRON AND MOSSBAUER SPECTROSCOPY 12 Hrs

PES : Theory of XPES, UVPES – evaluation of ionisation potential – chemical identification of elements – ESCA – Koopmann's theorem – chemical shift – UPES of NH₃, N₂, O₂, CO, H₂O and HCl – vibrational fine structure and their origin – evaluation of vibrational constants from UPES – spin-spin coupling - spin-orbit coupling.

Mössbauer spectroscopy: Discovery - Nuclei suitable for Mössbauer experiment - Hyperfine interactions - isomershifts quadrupole splitting & magnetic hyperfine splitting - temperature effects - applications of Mössbauer spectroscopy to compounds & complexes of Fe & Sn.

Text Book(s):

1. Ballhausen, C.J., Introduction to ligand field theory, Mc GrawHill, New York, 1962.
2. Figgis, B.N., Introduction to ligand fields, John Wiley & Sons Ltd., London and New York 1966.
3. Kettle, S.F.A., Coordination compounds, Thomas Nelson and Sons., Ltd., London, 1969.
4. Thomas M.Dunnand McClure and Pearson, "Some aspects of crystal theory fields, Harper-Row, New York, 1965.
5. Ballhanusen, C.J.,andHzry.B.Grey, "Molecular orbital theory" A Lecture note ReprintVolume,W. A. Benjamin, Inc. , New York, 1965.
6. Drago, R.S., Physical methods in chemistry,W.B.Saundar, 1977.
7. Cotton, F.A., and Wilkinson, Advanced Inorganic Chemistry,John Wiley & Sons Ltd.,London-New York, 1972.
8. Cotton, F.A., "Chemical Applications of Group Theory", Third Edition, Wiley Eastern Ltd., New Delhi, 2011.
9. Lever, A.B.P., Inorganic electronic Spectroscopy, Second Edition, Elsevier, Amsterdam, Oxford, New York, Tokio 1984.
10. Sutton,D., Electonic spectra of metal complexes, Second Edition McGraw-Hill, New York, 1968.
- 11.Figgis,B.N., Transition metal chemistry, volume IX, Marcel Uecker, Inc., Basel, 1982;
12. Hill, H. A. O., Day, P., Advanced Inorganic Chemistry, Darl H. McDaniel. J. Chem. Educ. , 1969
13. Ebsworth, E. A., David W. H. Rankin, Cradock, S.,Structural methods in Inorganic chemistry, ELBS.,CrcPr I Llc., 1991.
14. Purcell, K.F., Kotz J.C., Saunders W.B., Inorganic chemistry, Saunders College Publishing, Philadelphia, 1980.
15. Mabbs, F. E., Machin, D. J., Magnetism and transition metal complexes, John Wiley & Sons, Incorporated, 1973.

Reference Books:

1. Sutton, D., Electronic spectra of metal complexes, Second Edition, McGraw-Hill, New York, 1968.
2. Hill and Day, Advanced Inorganic Chemistry, First edition, John Wiley & Sons Ltd., 1968.
3. Walker, S., Straughan, B. P. Spectroscopy volume I, II & III, Chapman and Hall, 1976.
4. Figgis, B.N., Transition metal chemistry, volume IX, Marcel Uecker, Inc., Basel, 1982
5. Semen, A. A., Boris M.K., Electron spin resonance, Wiley, 1974
6. Rao, C.N.R., and Feraro, J R. Inorganic spectroscopy, volume I&II, Academic press, New York, 1971.
7. Carrington, A., McLachlan, A. D. Introduction to Magnetic Resonance, Harper and Row, New York, 1967.
8. Carrington, McLachlan A.D., Introduction to Magnetic Resonance: With Applications to Chemistry and Chemical Physics, Harper & Row, New York, 1979.
9. Boudreaux, F.A., and Muley, L.N., Theory and application of molecular paramagnetism, John Wiley & Sons, Inc., New York, 1976
10. Nakamoto, Infrared and Raman Spectra of Inorganic and Coordination Compounds, Wiley-Interscience, Fourth edition, 1998.
11. Vertes, A., and Burger, K., Mossbauer Spectroscopy, Elsevier, Amsterdam, 1979.
12. Gibbs, T.C., Mossbauer Spectroscopy, Chapman and Hall, London, 1977.
9. F.A. Boudreaux & L.N. Muley: Theory and application of molecular paramagnetism. John – Wiley
10. Nakamoto : Infrared and Raman Spectra of Inorganic & Coordination Compounds Wiley - Interscience IV Edn.
11. A. Vertes & K. Burger Mossbauer Spectroscopy Elsevier
12. T.C. Gibbs Mossbauer Spectroscopy Chapman - Hall